## (FILE 'HOME' ENTERED AT 13:05:19 ON 28 JUN 2005)

	FILE 'CAPLU	JS	' ENTERED AT 13:06:06 ON 28 JUN 2005
L1	186	S	PAROXETINE (2A) HYDROCHLORIDE
L2	46	S	PAROXETINE (2A) HCL
L3	192	s	L1 OR L2
L4	568116	S	MELT OR FUSION
L5	6	S	L3 AND L4
L6	608	s	MOLECULAR DISPERSION
L7	33	S	L6 AND L4
L8	0	S	L7 AND L3
L9	338778	S	DISPERSION
L10	10	S	L3 AND L9
L11	1	S	L10 AND L4

ANSWER 4 OF 6 CAPLUS COPYRIGHT 2005 ACS on STN

1999:42579 CAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 130:86197

Novel process for manufacturing paroxetine solid TITLE:

Krape, Philip J.; Chang, Sou-chan; Hein, William A., INVENTOR(S):

II; Teleha, Christopher A.

PATENT ASSIGNEE(S): Endo Pharmaceuticals Inc., USA

SOURCE: PCT Int. Appl., 32 pp.

CODEN: PIXXD2

DOCUMENT TYPE: LANGUAGE:

Patent English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

					APPLICATION NO.												
								WO 1998-US13350									
	W:	AL,	AM,	ΑT,	AU,	ΑZ,	BA,	BB,	BG,	BR	, BY,	CA,	CH,	CN,	CU,	CZ,	DE,
		DK,	EE,	ES,	FI,	GB,	GE,	GH,	GM,	GW.	, HU,	ID,	IL,	IS,	JP,	KΕ,	KG,
		ΚP,	KR,	ΚZ,	LC,	LK,	LR,	LS,	LT,	LU,	, LV,	MD,	MG,	MK,	MN,	MW,	MX,
		NO,	NZ,	PL,	PT,	RO,	RU,	SD,	SE,	SG	, SI,	SK,	SL,	ТJ,	TM,	TR,	TT,
		UA,	ŪĠ,	US,	UΖ,	VN,	YU,	ZW,	AM,	AZ,	, BY,	KG,	ΚZ,	MD,	RU,	ТJ,	TM
	RW:	GH,	GM,	KΕ,	LS,	MW;	SD,	SZ,	UG,	ZW	, AT,	ΒE,	CH,	CY,	DE,	DK,	ES,
		FI,	FR,	GB,	GR,	ΙE,	IT,	LU,	MC,	NL	, PT,	SE,	BF,	ВJ,	CF,	CG,	CI,
							ΝE,										
US	5955475				Α		1999	0921	US 1997-885068					19970630			
ZA	9805	488			Α		1999	0422		ZA :	1998-	5488			1	9980	624
AU	9881	717			A1		1999	0119		AU :	1998-	8171	7		1	9980	626
AU	733194				B2	2 20010510											
								EP 1998-931648				19980626					
EP	9914	8 0			В1		2005	0309									
	R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR.	, IT,	LI,	LU,	NL,	SE,	MC,	PT,
			FΙ														
BR	BR 9810231 TR 9903315			Α						1998-1							
TR	9903	315			T2		2000	0921	TR 1999-9903315					19980626			
	NZ 502062					2001	1026	NZ 1998-502062					19980626				
JP	JP 2002514227				T2		2002				1999-						
RU	U 2185820				C2		2002	0727	RU 2000-102637					19980626			
CN	1121						2003	0917	(	CN :	1998-	8079	66		1	9980	626
CA	2295	752			С		2004	1026	(	CA :	1998-:	2295	752		1	9980	626
CA	2295	752			AA		1999										
AT	2903	81			E		2005	0315		AT :	1998-	9316	48		1	9980	626
NO	9906	484					2000				1999-						
HK	1029	529			A1		2004	0102	]	HK 2	2001-	1003	18		2	0010	112
RIORIT	Y APP	LN.									1997-						
									1	WO :	1998-1	JS13	350	Ţ	W 1	9980	626

Solid dispersions of poorly soluble drugs are disclosed which are prepared AB using a solvent or fusion process. Such dispersions are manufactured with the free base of the drug, specifically paroxetine free base. Paroxetine free base and PEG-8000 were mixed and a stream of HCl gas was introduced to the mixture The solidified product was collected and 1H-NMR anal. showed the product was a mixture of PEG and paroxetine

·HCl. A tablet containing 22.21 mg paroxetine

·HCl was formulated.

THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT AB Solid dispersions of poorly soluble drugs are disclosed which are prepared

## 10019049

IT

using a solvent or fusion process. Such dispersions are manufactured with the free base of the drug, specifically paroxetine free base. Paroxetine free base and PEG-8000 were mixed and a stream of HCl gas was introduced to the mixture The solidified product was collected and 1H-NMR anal. showed the product was a mixture of PEG and paroxetine ·HCl. A tablet containing 22.21 mg paroxetine ·HCl was formulated. Mental disorder (depression, treatment of; manufacture of paroxetine solid dispersions using solvents or fusion process) Polyoxyalkylenes, uses RL: MOA (Modifier or additive use); USES (Uses)

IT

(manufacture of paroxetine solid dispersions using solvents or fusion process)

IT Drug delivery systems

> (tablets; manufacture of paroxetine solid dispersions using solvents or fusion process)

IT 9003-11-6, Ethylene glycol-propylene glycol copolymer 9003-39-8, PVP 9004-64-2, Hydroxypropyl cellulose 9004-65-3, Hydroxypropyl methyl cellulose 9004-67-5, Methyl cellulose 25322-68-3, Polyethylene glycol RL: MOA (Modifier or additive use); USES (Uses) (manufacture of paroxetine solid dispersions using solvents or fusion process)

ΙT 64-17-5, Ethanol, uses 67-56-1, Methanol, uses 67-63-0, Isopropanol, 71-23-8, Propanol, uses 71-36-3, Butanol, uses 78-83-1, Isobutanol, uses 78-92-2, sec-Butanol

RL: NUU (Other use, unclassified); USES (Uses) (manufacture of paroxetine solid dispersions using solvents or fusion process)

ΙT 61869-08-7, Paroxetine

RL: PEP (Physical, engineering or chemical process); PROC (Process) (manufacture of paroxetine solid dispersions using solvents or fusion process)

IT 78246-49-8P, Paroxetine hydrochloride

> RL: PNU (Preparation, unclassified); PRP (Properties); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses) (manufacture of paroxetine solid dispersions using solvents or fusion proce

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ANSWER 5 OF 6 CAPLUS COPYRIGHT 2005 ACS on STN
ACCESSION NUMBER:
                      1996:705783 CAPLUS
DOCUMENT NUMBER:
                       125:339069
                        Homogeneous mixtures of low temperature-melting drugs
TITLE:
                        and additives for controlled release
INVENTOR (S):
                        Cheskin, Howard; Hale, Thomas J.; Van Scoik, Kurt G.;
                        Zhou, Ji
PATENT ASSIGNEE(S):
                        Abbott Laboratories, USA
                        PCT Int. Appl., 23 pp.
SOURCE:
                        CODEN: PIXXD2
DOCUMENT TYPE:
                        Patent
                        English
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                       DATE APPLICATION NO.
    PATENT NO.
                      KIND DATE
                                                               DATE
                                         -----
    WO 9631197
                        A1 19961010 WO 1996-US4513
                                                               19960402
        W: CA, JP, MX
        RW: AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE
                        AA 19961010 CA 1996-2216934 19960402
    CA 2216934
                             19980121
                                         EP 1996-910705
                                                                19960402
    EP 818990
                        A1
        R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, PT, IE, FI
     JP 11503163 T2 19990323 JP 1996-530428 19960402
                                         US 1997-879468
    US 5807574
                              19980915
                                                                19970620
                        Α
                                                            A 19950403
W 19960402
PRIORITY APPLN. INFO.:
                                          US 1995-415401
                                          WO 1996-US4513
    Disclosed herein is a controlled-release formulation comprising, in
AB
     combination a therapeutically-effective dosage of drug which melts
     at low temperature and an additive selected from the group consisting of Et
     cellulose, Me cellulose, hydroxypropyl cellulose, polyacrylamide,
     ethylene-vinyl acetate copolymer, poly(Me methacrylate), polyhydroxyethyl
    methacrylate and waxes, and the like, such that the additive and the drug
     form a homogeneous drug-additive composite, wherein the drug is selected
     from the group consisting of: divalproex sodium (I), ibuprofen, ramipril,
     dibenzyline, erythrityl tetranitrate, isosorbide dinitrate, methsuximide,
    ketoprofen, gemfibrozil, paroxetine HCl, and
     trimipramine maleate. I 25 g were melted with 1.25 g of polyethylene wax
    at 115° and the resulting molten composite was filled into
     capsules, which were subjected to USP dissoln. test in simulated gastric
     fluid. The capsules showed controlled release of I such that only
     .apprx.60 % of the I in the composite was released over 24 h.
    Disclosed herein is a controlled-release formulation comprising, in
AB
    combination a therapeutically-effective dosage of drug which melts
     at low temperature and an additive selected from the group consisting of Et
     cellulose, Me cellulose, hydroxypropyl cellulose, polyacrylamide,
     ethylene-vinyl acetate copolymer, poly(Me methacrylate), polyhydroxyethyl
    methacrylate and waxes, and the like, such that the additive and the drug
     form a homogeneous drug-additive composite, wherein the drug is selected
     from the group consisting of: divalproex sodium (I), ibuprofen, ramipril,
     dibenzyline, erythrityl tetranitrate, isosorbide dinitrate, methsuximide,
    ketoprofen, gemfibrozil, paroxetine·HCl, and
     trimipramine maleate. I 25 g were melted with 1.25 g of polyethylene wax
     at 115° and the resulting molten composite was filled into
    capsules, which were subjected to USP dissoln. test in simulated gastric
     fluid. The capsules showed controlled release of I such that only
     .apprx.60 % of the I in the composite was released over 24 h.
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59-96-1, Dibenzyline 77-41-8, Methsuximide 87-33-2, Isosorbide

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dinitrate 521-78-8, Trimipramine maleate 7297-25-8, Erythrityl tetranitrate 9003-05-8, Polyacrylamide 9004-57-3, Ethyl cellulose 9004-64-2, Hydroxypropyl cellulose 9004-67-5, Methyl cellulose 9011-14-7, Polymethylmethacrylate 15687-27-1, Ibuprofen 22071-15-4, Ketoprofen 24937-78-8, Ethylene-vinyl acetate copolymer 25249-16-5 25812-30-0, Gemfibrozil 76584-70-8, Divalproex sodium 78246-49-8, Paroxetine hydrochloride 87333-19-5, Ramipril RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (homogeneous mixts. of low temperature-melting drugs and additives for controlled release)

ANSWER 6 OF 6 CAPLUS COPYRIGHT 2005 ACS on STN ACCESSION NUMBER: 1988:192668 CAPLUS DOCUMENT NUMBER: 108:192668 Solid-state forms of paroxetine TITLE: hydrochloride AUTHOR (S): Buxton, P. Christopher; Lynch, Ian R.; Roe, John M. Biosci. Res. Cent., Beecham Pharm., Epsom Surrey, KT18 CORPORATE SOURCE: 5XO, UK SOURCE: International Journal of Pharmaceutics (1988), 42(1-3), 135-43 CODEN: IJPHDE; ISSN: 0378-5173 DOCUMENT TYPE: Journal LANGUAGE: English Paroxetine-HCl (I) exists in two solid state forms differentiated by their degrees of hydration. Form I is a non-hygroscopic hemihydrate and is thermodynamically the more stable. Form II is a hygroscopic anhydrate the moisture content of which is controlled by the prevailing humidity. Form II converts to Form I, if seed crystals of Form I are present, when exposed to humid conditions or if subjected to compression. The rates of transformation were determined by IR spectroscopy and techniques are described to identify the solid state form in compressed tablets. The transformation follows kinetic models described by diffusion and phase boundary processes and the rate constant (k) is related to temperature by the Arrhenius equation. At constant temperature ln k is related to the reciprocal of the compaction pressure. Thermodn. measurements of free energy ( $\Delta GT$ ) and enthalpy ( $\Delta HT$ ) show the two forms to be energetically similar and measurements of dissoln. indicate that both forms would be expected to be bioequivalent. Solid-state forms of paroxetine hydrochloride TТ Paroxetine-HCl (I) exists in two solid state forms AB differentiated by their degrees of hydration. Form I is a non-hygroscopic hemihydrate and is thermodynamically the more stable. Form II is a hygroscopic anhydrate the moisture content of which is controlled by the prevailing humidity. Form II converts to Form I, if seed crystals of Form I are present, when exposed to humid conditions or if subjected to compression. The rates of transformation were determined by IR spectroscopy and techniques are described to identify the solid state form in compressed tablets. The transformation follows kinetic models described by diffusion and phase boundary processes and the rate constant (k) is related to temperature by the Arrhenius equation. At constant temperature ln k is related to the reciprocal of the compaction pressure. Thermodn. measurements of free energy ( $\Delta GT$ ) and enthalpy ( $\Delta HT$ ) show the two forms to be energetically similar and measurements of dissoln. indicate that both forms would be expected to be bioequivalent. Heat of fusion and Heat of freezing IT Heat of solution (of paroxetin hydrochloride solid-state forms) IT Solution rate (of paroxetine hydrochloride solid-state forms) IT Free energy (of transition, of paroxetine hydrochloride solid-state forms) Compaction TT

(paroxetine hydrochloride solid-state forms in

Humidity

relation to)

## 10019049

IT 78246-49-8, Paroxetine hydrochloride RL: PRP (Properties) (solid-state forms of)